## ON THE STEREOCHEMICAL ASPECTS OF THE 1,1-CYCLOADDITION REACTION OF DIAZOALKENES

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Abstract: Allyl substituted diazoalkenes undergo intramolecular 1,1-cycloaddition with complete retention of configuration to give 1,2-diazabicyclo[3.1.0]hex-2-enes.

The additions of diazoalkanes to olefins are amongst the most thoroughly studied 1,3dipolar cycloadditions.<sup>1-4</sup> The reactions of simple diazoalkanes are HO(1,3-dipole)-LU(dipolarophile) controlled.<sup>5,6</sup> 3-Substituted pyrazolines are the major products produced; a result of union of the larger diazoalkane HO coefficient on carbon with that of the larger dipolarophile LU coefficient on the unsubstituted carbon.<sup>5</sup> Recent work from our laboratory<sup>7</sup> as well as studies by Miyashi and Mukai<sup>8</sup> has shown that intramolecular 1,1-cycloaddition of diazoalkenes can compete with the normal 1,3-addition mode when certain geometric constraints are imposed.<sup>9</sup> Thus, various allyl substituted diazomethanes have been found to undergo a formal nitrene type 1,1-cycloaddition to give 1,2-diazabicyclo[3.1.0]hex-2-enes on heating.<sup>7,8</sup> In this communication we wish to describe the stereochemical course associated with this novel cycloaddition reaction.<sup>10</sup>



Thermolysis of the sodium salt of <u>E</u>-1,4-diphenyl-3-buten-1-one N-tosyl hydrazone (<u>1</u>) at  $80^{\circ}$ C in benzene for 20 min produced the corresponding diazo compound <u>5</u> which cyclized with complete stereospecificity to give <u>exo</u>-3,6-diphenyl-1,2-diazabicyclo[3.1.0]hexene (<u>6</u>).<sup>7</sup> In order to probe the stereochemical course of the l,l-cycloaddition reaction, we studied the

thermolysis of the related <u>Z</u>-stereoisomer  $\underline{2}$ . Heating the sodium salt of  $\underline{2}$  at  $80^{\circ}$  for 20 min resulted in the formation of the <u>cis</u>-diazoalkene; ir  $2000 \text{cm}^{-1}$ ; NMR (CDCl<sub>3</sub>,60 MHz)  $\delta$  3.40 (dd, 2H, J=7.0 and 2.0 Hz), 5.80 (<u>dt</u>, 1H, J=11.0 and 7.0 Hz), 6.80-7.50 (<u>m</u>, 11H). Unfortunately, we were unable to isolate any characterizable material from the further reaction of this diazoalkene.



This problem was circumvented by studying the cycloaddition behavior of the tosylhydrazones of <u>E-(3)</u> and <u>Z</u>-5-phenylpent-2-en-5-one (<u>4</u>). Heating a benzene solution of the sodium salt of the <u>E</u>-isomer <u>3</u> at 80<sup>°</sup> for 1 hr followed by allowing the mixture to stand at room temperature for 48 hr gave <u>exo</u>-3-phenyl-6-methyl-1,2-diazabicyclo[3.1.0]hex-2-ene (<u>7</u>), mp 53-54<sup>°</sup>, in 53% isolated yield as the exclusive 1,1-cycloadduct; NMR (CDCl<sub>3</sub>,400 MHz) & 1.36 (<u>d</u>, 3H, J=6.0 Hz), 1.52 (<u>dq</u>, 1H, J=6.0 and 4.0 Hz), 2.74 (<u>ddd</u>, 1H, J=6.0, 4.0 and 3.0 Hz), 3.36 (<u>dd</u>, 1H, J=18.0 and 6.0 Hz), 3.44 (<u>dd</u>, 1H, J=18.0 and 3.0 Hz) and 7.2-8.1 (<u>m</u>, 5H); C<sup>13</sup> NMR (CDCl<sub>3</sub>) 16.9 (CH<sub>3</sub>), 38.0 (C<sub>4</sub>), 46.7 (C<sub>5</sub>), 48.8 (C<sub>6</sub>), 168.3 (C<sub>3</sub>); uv (95% ethanol) 255 nm ( $\epsilon$  12100); m/e 172 (M<sup>+</sup>), 144, 129 (base), 128 and 103. When the sodium salt of the Z-tosylhydrazone <u>4</u> was subjected to similar reaction conditions, the major product obtained was endo-3-phenyl-6-methyl-1,2-diazabicyclo[3.1.0]hex-2-ene (<u>8</u>). NMR analysis of the crude reaction mixture showed that the corresponding <u>exo</u> isomer  $\underline{7}$  was present in less than 5% yield. The structure of the endo isomer was based on its characteristic 90-MHz NMR spectrum (CDCl<sub>3</sub>) & 0.98 (<u>d</u>, 3H, J=6.0 Hz), 2.42-2.90 (<u>m</u>, 2H), 3.01 (<u>dd</u>, 1H, J=18.0 and 3.0 Hz), 3.20 (<u>dd</u>, 1H, J=18.0 and 8.0 Hz), and 7.2-7.8 (<u>m</u>, 5H) and was further supported by its C<sup>13</sup> NMR (CDCl<sub>3</sub>) 5.7 (CH<sub>3</sub>), 34.3 (C<sub>4</sub>), 41.6 (C<sub>5</sub>), and 42.2 (C<sub>6</sub>).

The above experimental observations clearly indicate that the intramolecular 1,1-cycloaddition reaction of allyl substituted diazoalkenes is highly stereospecific. Reasonable mechanistic options for the 1,1-cycloaddition include a concerted pathway or a stepwise process. The concerted path would parallel the stereospecific addition of singlet nitrenes to olefins which proceeds with retention of stereochemistry about the double bond.<sup>11</sup> Stepwise nucleophilic attack of the terminal double bond on the electron deficient nitrogen atom of diazoalkene <u>5</u> can generate a six-membered ring dipole which contains an aza-allylic carbanion as well as a secondary carbonium ion. Collapse of this new 1,3-dipole will generate the diazabicyclo hexane ring system. As long as the closure step is fast relative to ring flip-



ping, the l,l-cycloaddition will proceed with retention of configuration. The stereospecificity of the above reaction is similar to that involved in the intramolecular l,l-cycloaddition of nitrile ylides  $^{12,13}$  and nitrile imines.  $^{14,15}$ 

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- R. Huisgen, R. Grashey and J. Sauer in "The Chemistry of Alkenes", S. Patai, Ed., Interscience, London, 1964, pp 806-878.
- 2. G. W. Cowell and A. Ledwith, Quart. Rev., <u>24</u>, 119 (1970).
- R. Huisgen, J. Koszinowski, A. Ohta and R. Schiffer, <u>Angew. Chem. Int. Ed. Engl.</u>, <u>19</u>, 202 (1980).
- 4. R. A. Firestone, Tetrahedron Lett., 2209 (1980).
- 5. K. N. Houk, J. Sims, C. R. Watts and L. J. Luskus, <u>J. Am. Chem. Soc.</u>, <u>95</u>, 7301 (1973).
- 6. R. Sustmann, Tetrahedron Lett., 2717 (1971).
- 7. A. Padwa and H. Ku, Tetrahedron Lett., 1009 (1980).
- 8. Y. Nishizawa, T. Miyashi and T. Mukai, <u>J. Am. Chem. Soc.</u>, <u>102</u>, 1176 (1980).
- 9. For a recent review on intramolecular 1,3-dipolar cycloaddtitions, see A. Padwa, New Synthetic Methods, Vol. 5, Verlag Chemie, N. Y., N. Y., pp 25-69, 1979.
- 10. We have recently been informed by Dr. Miyashi and Mukai that they have obtained identical results. We wish to thank these authors for a preprint of their work.
- 11. W. Lwowski, Nitrenes, Interscience Publ. N. Y., N. Y., 1970.
- 12. A. Padwa and P. H. J. Carlsen, J. Am. Chem. Soc., <u>99</u>, 1514 (1977).
- 13. J. Fischer and W. Steglich, Angew. Chem. Int. Ed. Engl., <u>18</u>, 167 (1979).
- 14. L. Garanti and G. Zecchi, J. Chem. Soc. Perkin Trans. 1, 2092 (1977).
- A. Padwa and S. Nahm, <u>J. Org. Chem.</u>, <u>44</u>, 4746 (1979).
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